A New Pigment from Tegula funebralis

(Mollusca: Gastropoda)

BY

PATRICIA MCGEE

Hopkins Marine Station of Stanford University, Pacific Grove, California

(1 Text figure; 3 Tables)

IN THE TROCHACEAN SPECIES Tegula funebralis (A. ADAMS, 1854) and T. brunnea (PHILIPPI, 1848) which are abundant in the intertidal zone of the Pacific coast, the female gonad is bright green. In both species, the pigment is found in droplets evenly dispersed throughout the yolk. An extraction of the pigment in T. funebralis was made. The crude green pigment was partitioned into a group of yellow carotenoids and an unknown green pigment. This is a study of these colored materials.

PREPARATION

The pigment was initially extracted from eggs carefully stripped from 200 gonads. The eggs were blended in a Waring blender with absolute methanol for five minutes. The methanol was changed, and extraction was repeated until the suspension was white. The crude green pigment was then dried and redissolved in methanol. This reextraction was repeated three times. Ether and water partition of the pigment separated the material into a yellow epiphase and a green hypophase. Repeated partitioning was used to purify the materials.

RESULTS

The yellow material, dried and dissolved in petroleum ether (Bp. 40° 60° C) was placed on an Al_2O_a column. The column was developed with a gradient of acetone in petroleum ether. The four bands observed were collected, and a tentative identification of zeaxanthin, lutein and alpha carotene was made from the data in Table I (P. KARRER & E. JUCKER, 1950).

The green pigment could not be identified. Some of its properties are briefly stated in Table II. In addition, in aqueous solution it freely passed through sephadex G-75 which indicates a molecular size larger than that corresponding to a molecular weight of 40,000. In aqueous solution, it may be warmed to 100° C without obvious change. Prolonged heating at 100° C resulted in a brown discoloration, and is assumed to be due to decomposition. It is photosensitive, and becomes yellow with prolonged exposure to light. When reduced with hydrosulfite, a yellow color appears, but reoxidation to green can be achieved by autoxidation or treatment with H_2O_2 . In methanol, a green fluorescence was observed in the oxidized form.

The large molecular size suggested a protein complex. The unknown was, therefore, placed in aqueous solution and an equal amount of CHC1₃ with .1 volume amylalcohol added. This was vigorously shaken and centrifuged for 10 minutes. A blue protein appeared between the phases of green aqueous solution and colorless CHC1₃. The aqueous phase was re-treated until the blue zone no longer appeared. The solubilities and spectrum of the un-

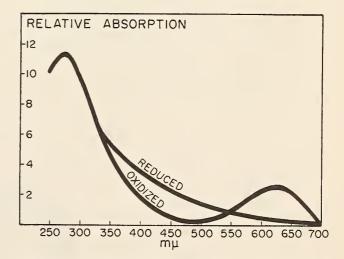


Figure 1: The absorption maxima of the unknown green pigment in H₂O. The oxidized state peaks at 640 and 273, while the reduced state peaks at 273μ .

Page	26
------	----

BAND	Absorption Maxima in CS ₂	REMARKS	CAROTENOID*
Ι	517 482 450	hypophasic in petroleum ether and 90% MeOH	zeaxanthin
II	510 470 442	distributed in both phases of pet. ether and 90% MeOH	
III	508 475 445	hypophasic in petroleum ether and 90% MeOH	lutein
IV	509 477	epiphasic in petroleum ether and 90% MeOH	α - carotene

Table 1 **PROPERTIES OF YELLOW PIGMENTS**

absorption spectra and solubility properties.

PIGMENT	SOLUBILITY	ABSORPTION	MAXIMA mµ	COLOF	ATION
		oxidized	reduced	oxidized	reduced
1. Native	s. H2O, MeOH, EtOH, acet. i. eth., CHC13 pet. eth., CS2	H2O 640 273 MeOH 640 370*(s)	H₂O 273	neutral- green acid- yellow alkaline- yellow with ppt.	neutral- yellow acid- yellow alkaline- yellow
2. Deproteinized	same as 1.	same as 1	same as 1	same as 1 except alkaline- yellow with no ppt.	same as 1
3. Allagochrome (Habermann, 1960)	s. H2O	H₂O 630 320 260	H₂O 320 260	neutral- green acid- red alkaline- brown	neutral- yellow- orange acid- ? alkaline- ?
* (s) ==	shoulder.	i.	= soluble = insoluble = ether		

Table 2			
PROPERTIES	OF	GREEN	PIGMENT

eth. = ether

pet. eth. == petroleum ether

acet. = acetone

Table 3

Results of mass speetral analysis of erude green pigment in percent by weight

ELEMENT	PERCENT	ELEMENT	PERCENT
Al	0.005	Na	1.0
Ca	0.10	Pb	0.01
Cu	0.005	Si	0.005
Fe	0.01	Sn	0.01
Mg	0.03	Zn	trace
Mn	trace		

known without protein appear in Table II. It should be noted that the absorption spectrum of the oxidized form was not changed.

In preliminary observations of the development of the eloscly related *Tegula brunnea* the green pigment present in the eggs was observed to eoncentrate in the eiliated cells of the trochophore. The amount appeared to inerease as the velum formed.

DISCUSSION

The unknown pigment resembles Allagochrome (HABER-MANN, 1960; GARRICK & HABERMANN, 1962) in its absorption spectrum and oxidation-reduction activities. A comparison of the two pigments' properties can be seen in Table II. Allagochrome is present in a variety of higher plants and its function is hypothesized to be respiratory due to the ease with which oxidation and reduction can be induced.

The peak at 273 m μ also is suspiciously near the characteristic peak, 275 m μ of the coenzyme Q, a lipid soluble quinone (CRANE, 1959). There is a broad distribution of the five known forms of coenzyme Q in aerobic tissues. It has been found in all vertebrates, higher plants, aerobic baeteria, invertebrates and red and green algæ. The view is that eoenzyme Q is necessary for respiratory electron transfer occurring in the mitochondria and equivalent structures. While several quinones of biological origin have been described, their actual involvement in electron transport has seldom been demonstrated. If a respiratory function for the unknown pigment is postulated, the observed migration of the pigment to the ciliated cells of the velum of the trochophore and veliger larvæ may be of signifieance. These cells would be expected to have a high metabolic activity.

SUMMARY

The pigments of the eggs of *Tegula funebralis* were extracted in methanol. This crude green pigment contained three yellow materials with spectral properties resembling the carotenoids: zeaxanthin, lutein and alpha earotene, and an unknown green pigment. The green pigment was found to have an attached protein, absorption maxima in H₂O at 640, 273m μ in the oxidized form and 273 m μ in the reduced state, and a marked resemblance to known quinones that have been suggested to act as respiratory pigments.

LITERATURE CITED

CRANE, F. L.

1959. Internal distribution of Coenzyme Q in higher plants. Plant Physiol. 34: 128-131

GARRICK, L. S. & H. M. HABERMANN

1962. Distribution of allagochrome in vascular plants. Amer. Journ. Bot. 49: 1078 - 1088

HABERMANN, H. M.

KARRER, P. & E. JUCKER

1950. Caretonoids. Elsevier Publ. Co., Inc. New York etc. $x\,+\,384$ pp.



^{1960.} A new leaf pigment (pp. 73-82 in:) Comp. Biochem. of photoreactive systems; Acad. Press, New York and London. xii + 437 pp.